

A NEW APPROACH TO THE TOTAL SYNTHESIS OF STEROIDS

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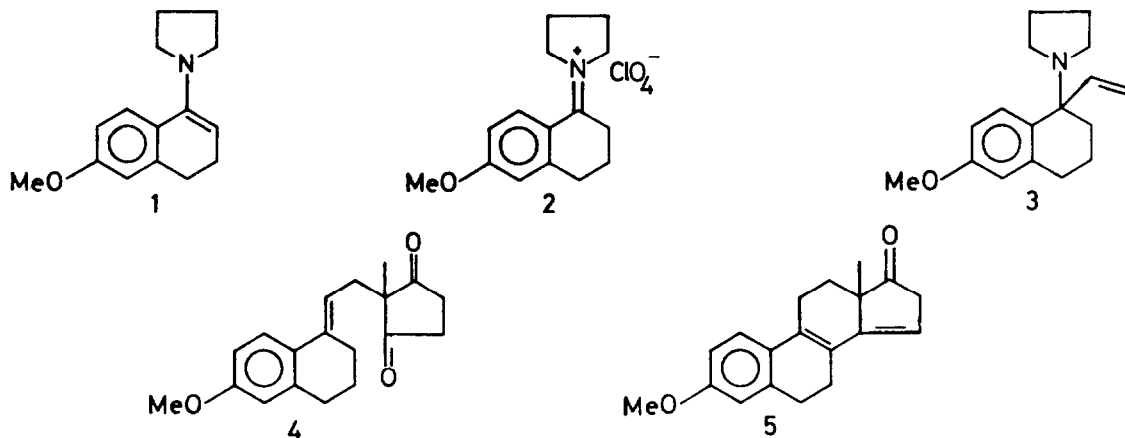
The simple three-step conversion of 6-methoxytetralone-1 to $\Delta^{8,14}$ -steroids (5), initially developed by Torgov et.al.¹, has proved to be a versatile approach for the synthesis of natural estrogenic hormones or their ring A or B modified analogues². In connection with our studies on the enamines derived from α -tetralones we have found that the corresponding iminium salts serve as valuable precursors for a convenient alternate route to 5. This communication presents the transformation of enamine 1 to 8,14-bis-dehydroestrone methyl ether (5).

1-Pyrrolidyl-6-methoxy-3,4-dihydronaphthalene (1) was obtained from 6-methoxytetralone-1, in 80 p.c. yield, by the method of White and Weingarten³. Its structure followed from its IR ($\nu_{\text{max}}^{\text{neat}}$ 1615, N=C=C) and NMR (δ^{CDCl_3} 5.01, t, 1H, C=C-H, J=5 c/s). Treatment of 1 with HClO_4 in glacial acetic acid gave crystalline perchlorate 2 (80-90 p.c.), which could be stored under nitrogen without deterioration.

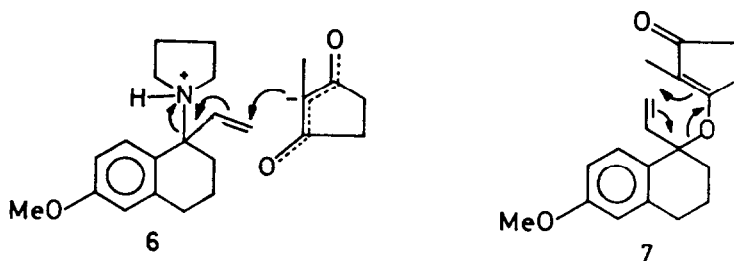
Addition of 2 to a ten-fold excess of vinylmagnesium bromide, in tetrahydrofuran, resulted in the formation of allyl amine 3, as an oil, in 82 p.c. yield (based upon recovered tetralone). Structure of 3 was attested by its spectral data and confirmed by conversion of the oily amine to its picrate, m.p. 103-104°. Found : C, 56.81; H, 5.49; N, 11.40. $\text{C}_{23}\text{H}_{26}\text{N}_4\text{O}_8$ requires : C, 56.78; H, 5.39; N, 11.52.

When a mixture of 2-methylcyclopentanedione-1,3 and allyl amine 3, in methanol, was refluxed for 16 hours, the seco-steroid 4 was formed in 80 p.c. yield. While no effort was spent on isolating 4 in its crystalline form, the IR and NMR spectra of the product were identical to those of a sample of 4 prepared according to the procedure of Ananchenko and Torgov¹. Conventional acid-catalyzed cyclization of 4 gave the known 8,14-bisdehydroestrone methyl ether 5, m.p. 108-110°.

The condensation of methylcyclopentanedione with allyl amine 3 may follow the mechanism suggested for the analogous reaction with the corresponding allyl alcohol⁴. In the present case this would involve protonation of the amine by the di-



ketone, acting as an acid, to give presumably the ion-pair system 6, which collapses by an SN_2 ' displacement of pyrrolidine. An alternate mechanism⁵ in which allyl ether 7 - initially formed from 3 and methylcyclopentanedione - undergoes a sigma-tropic rearrangement under conditions of the reaction, to form 4, cannot, however, be excluded.



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References :

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